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CHEMISTRY I (TESC 141) STUDY GUIDE MOLES/ STOICHIOMETRY Mole—a unit of measurement that expresses the amount of atoms, molecules or some other unit. The number of items in one mole is commonly referred to Avogadro's number which equals 6.022×10^{23} .

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Monday (1/27): In-class: Naming, Formula, and Balancing Review; HW: Finish Review, Watch Stoichiometry Lecture Video and Complete Guided Notes (p 1-3) Due Tomorrow Tuesday (1/28): In-class: Note Check, Funsheets A-E; HW: Study for Quiz Friday Wednesday (1/29): In-class: Funsheets A-E; HW: Watch Percent Yield and Limiting Reactant Lecture Video and Complete Guided Notes (p 4-6) Due Monday ...

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Unit 10: Stoichiometry - Chemistry by Mrs. Hinkson

Title: Unit 8 Stoichiometry Study Guide Answers Author: media.ctsnet.org-Julia Eichmann-2020-10-01-06-16-48 Subject: Unit 8 Stoichiometry Study Guide Answers

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TEACHER GUIDE AND ANSWERS Study Guide - Chapter 11 - Stoichiometry Section 11.1 What is stoichiometry?
1. true 2. true 3. false 4. true 5. true 6. 2, 2, 64.10 7. 3, 3, 96.00 8. 2, 2, 88.02 9. 4, 4, 72.08 10. methanol and oxygen gas 11. carbon dioxide and water 12. 160.10 g 13. 160.10 g 14. They are equal. 15. A mole ratio is a ratio between ...

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