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Key

Naming Ionic Compounds – Answer Key Give the name of the following ionic compounds: Name 1) Na_2CO_3 sodium carbonate 2) NaOH sodium hydroxide 3) MgBr_2 magnesium bromide 4) KCl potassium chloride 5) FeCl_2 iron (II) chloride 6) FeCl_3 iron (III) chloride 7) $\text{Zn}(\text{OH})_2$ zinc hydroxide 8) BeSO_4 beryllium sulfate 9) CrF

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Naming Ionic Compounds Directions: Name the following compounds using the Stock Naming System.

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1. CaCO_3 calcium carbonate_____ 2. KCl potassium chloride_____ 3. FeSO_4 iron (II) sulfate_____ 4. LiBr lithium bromide_____ 5. MgCl_2 magnesium chloride_____ 6. FeCl_3 iron (III) chloride_____ 7. $\text{Zn}_3(\text{PO}_4)_2$

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Answers – Naming Chemical Compounds 1) NaBr sodium bromide 2) $\text{Ca}(\text{C}_2\text{H}_3\text{O}_2)_2$ calcium acetate 3) P_2O_5 diphosphorus pentoxide 4) $\text{Ti}(\text{SO}_4)_2$ titanium(IV) sulfate 5) FePO_4 iron(III) phosphate 6) K_3N potassium nitride 7) SO_2 sulfur dioxide 8) CuOH copper(I) hydroxide 9) $\text{Zn}(\text{NO}_2)_2$ zinc nitrite 10) V_2S_3 vanadium(III) sulfide

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Mrs. Primatesta

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Key Points. Most cations and anions can combine to form neutral, solid compounds that are usually known as salts. The net charge of an ionic compound must be zero. Therefore, the number of cations and anions in an

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ionic compound must be balanced to make an electrically neutral molecule. When naming ionic compounds, the cation retains the same name as the element.

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Naming Ions And Chemical Compounds 5 Answer Key
Worksheets ...

Naming Ionic Compounds Recall that ionic compounds consist of a positively charged cation and a negatively charged anion. Ions (of either variety) may contain either a single element or more than one element. (When an ion consists of more... To recognize an ionic compound, look for the presence of a ...

Steps to Naming Ionic and Covalent Compounds -
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Name of Ion Formula of Ion Formula of Acid Suffix on
the Name of the Ion Name of Acid. acetate CH_3COO .

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1 – HCH. 3COO ate acetic acid borate BO₃ 3 – H. 3BO₃ate
boric acid bromate BrO₃ 1 – HBrO. 3ate bromic acid
bromide Br 1 – HBr ide hydrobromic acid bromite BrO₂
1 – HBrO. 2ite bromous acid carbonate CO₃ 2 – H.

Nomenclature #1: Binary Ionic Compounds

Practice naming ionic compounds when given the formula. Practice naming ionic compounds when given the formula. If you're seeing this message, it means we're having trouble loading external resources on our website. If you're behind a web filter, please make sure that the domains *.kastatic.org and *.kasandbox.org are unblocked.

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Ionic compounds are composed of positively and negatively charge ions held together by strong ELECTROSTATIC attraction. 8. A positively charged ion such as Ca^{2+} is called CATION and it is formed when an atom of calcium LOSES 2 electrons. The cation will have as many electrons as the noble gas ARGON. !

Ionic and Covalent Compounds Name: KEY
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