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Chapter revision

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Guaranteed Questions - Coordination Compounds Coordination Compounds Class  
12 | NCERT Book Tick Mark p14 | 12th Board | Arvind Sir ~~MCQ: Coordination~~  
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Coordination Compounds Multiple Choice Questions and Answers for competitive exams. These short objective type questions with answers are very important for Board exams as well as competitive exams like IIT-JEE, AIIMS etc. These short solved questions or quizzes are provided by Gkseries.

Coordination Compounds Multiple Choice Questions and ...

A24.6.7. a. There are 6 ligands bonded to the central atom, so the complex is octahedral. Since the complex has a negative 3 charge and the oxidation number of Cl is -1, Cl 6 gives the complex a -6 charge and Ti must be +3 for the complex to be -3 in total.. Ti is the second transition metal in the first row of transition metals (atomic number 22).

24.E: Chemistry of Coordination Chemistry (Exercises ...

Short Answer Type Questions Arrange the following complexes in the increasing order of conductivity of their solution:  $[\text{Co}(\text{NH}_3)_3\text{Cl}_3]$ ,  $[\text{Co}(\text{NH}_3)_4\text{Cl}_2]\text{Cl}$ ,  $[\text{Co}(\text{NH}_3)_6]\text{Cl}_3$ ,  $[\text{Cr}(\text{NH}_3)_5\text{Cl}]\text{Cl}_2$ ; A coordination compound  $\text{CrCl}_3 \cdot 4\text{H}_2\text{O}$  precipitates silver chloride when treated with silver nitrate. The molar conductance of its solution corresponds to a total of two ions.

Class 12 Important Questions for Chemistry - Coordination ...

Extensive lecture notes from the lectures Equity and Trust Law 2013/14 (64 pages)  
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Solved Examples of Coordination Compounds. Question 1: Write down the molecular formulae of the following co-ordination compounds. (i) Hexaammine iron (III) nitrate (ii) Ammonium tetrachlorocuprate (II) (iii) Sodium monochloropentacyanoferrate (III) (iv) Potassium hexafluorocobaltate (III) Solution: (i)  $[\text{Fe}(\text{NH}_3)_6](\text{NO}_3)_3$  (ii)  $(\text{NH}_4)_2[\text{CuCl}_4]$

Coordination Compounds – Solved Problems for IIT JEE ...

Online Chemistry Questions and Answers – Basic Chemistry Part 2 (26-50) 26) The largest number of molecules is in (a) 36 g of water (b) 28 g of carbon monoxide (c) 46 g of ethyl alcohol (d) 54 g of nitrogen pentoxide Answer: A. 27) What are

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isobars? Answer: Same mass number but a different atomic number.

100 Chemistry Questions and Answers - Basic Chemistry ...

In coordination chemistry, the donor atom of a ligand is (a) a Lewis acid. (b) the counter ion (c) the central metal atom. (d) the atom in the ligand that shares an electron pair with the metal. (e) the atom in the ligand that accepts a share in an electron pair from the metal. 3. Consider the coordination compound,  $\text{Na}_2 [\text{Pt}(\text{CN})_4]$ . The Lewis acid is

Sample Questions - Chapter 25 - Department of Chemistry

Very Short Answer Type Questions [1 Mark] Question 61: Give an example of coordination isomerism. Answer:  $[\text{Co}(\text{NH}_3)_6][\text{Cr}(\text{CN})_6]$  and  $[\text{Cr}(\text{NH}_3)_6][\text{Co}(\text{CN})_6]$  are coordination isomer. Question 62: Give an example of ionisation isomerism. Answer:  $[\text{Co}(\text{NH}_3)_5\text{Br}]\text{SO}_4$  and  $[\text{Co}(\text{NH}_3)_5\text{SO}_4]\text{Br}$  is the example of ionisation isomerism. Short Answer Type Questions [I] [2 Marks]

Important Questions for CBSE Class 12 Chemistry ...

9.27. Discuss briefly giving an example in each case the role of coordination compounds in : (a) biological systems, (b) analytical chemistry, (c) medicinal chemistry, and (d) extraction/metallurgy of metals. Ans: (i) Coordination compounds are of great importance in biological systems.

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NCERT Solutions For Class 12 Chemistry Chapter 9 ...

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i)  $c$  (speed of light) =  $\lambda$ (wavelength)  $\times$   $\nu$ (frequency) ii)  $c$  (m s<sup>-1</sup>) =  $\lambda$ (m)  $\times$   $\nu$ (Hz) [Hz = s<sup>-1</sup>] iii)  $\nu = c / \lambda$ . iv)  $\nu = c / \lambda = 3 \times 10^8 \text{ m s}^{-1}$  (speed of light) /  $456 \times 10^{-9} \text{ m}$  (wavelength provided) =  $6.58 \times 10^{14} \text{ Hz}$ . v)  $\lambda = c / \nu$ . vi)  $\lambda = 3 \times 10^8 \text{ m s}^{-1}$  (speed of light) /  $2.45 \times 10^9 \text{ s}^{-1}$  (frequency provided) = 0.122 m.

## FIRST YEAR UNDERGRADUATE CHEMISTRY

Question 9. Construct two complexions using Fe as the central metal ion and CN<sup>-</sup> as the ligands, such that one is paramagnetic and the other is diamagnetic Give their IUPAC name. Answer: [Fe (CN)<sub>6</sub>]<sup>3-</sup> Hexacyanoferrate (iii) ion is paramagnetic and d<sup>2</sup>sp<sup>3</sup> hybridisation. Question 10. IUPAC name of the coordination compound [Co(NH<sub>3</sub>)<sub>5</sub>Cl]Cl<sub>2</sub> is pentamminechlorido cobalt (III) chloride.

Plus Two Chemistry Chapter Wise Questions and Answers ...

Exam 4, MO theory and coordination compounds Chapter 9, end and Chapter 24. MO theory: Rules: □ 1. The number of MO 's equals the # of Atomic orbitals □ 2.

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The overlap of two atomic orbitals gives two molecular orbitals, 1 bonding, one antibonding □ 3. Atomic orbitals combine with other atomic orbitals of similar energy.

Chapter 24 Chemistry of Coordination Compounds

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