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Library of Congress Cataloging-in-Publication Data:  
Levenspiel, Octave. Chemical reaction engineering 1  
Octave Levenspiel. - 3rd ed. p. cm. Includes index.  
ISBN 0-471-25424-X(cloth : alk. paper) 1. Chemical  
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Octave Levenspiel was a professor of the field Chemical engineering at Oregon State University. In this vast and evergreen field his major interests lied in Chemical Reaction Engineering which is one of the core subjects in Chemical Engineering. He published this book which is considered as a Bible for understanding major concepts of Chemical Reaction Engineering.

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dependence on temperature We will once again look  
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and hydrogen (see section Chemical equilibrium of the thermodynamics chapter). This reaction follows the equation:  $N_2 + 3H_2 \rightleftharpoons 2NH_3$  (1)  $H_0 = 92 \text{ kJ mol}^{-1}$   $S_0 = 192 \text{ J mol}^{-1} \text{ K}^{-1}$  To find the Gibbs free energy of ...

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