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www.njctl.org Chemistry Atomic

Structure Answers 1. In the nuclear model of the atom, protons (and neutrons) are housed in a small, dense nucleus. Electrons surround the nucleus in an area of mostly empty space. 2. If electrons are electrically attracted to nucleus and would, therefore, have centripetal acceleration in order to orbit the nucleus.

Atomic Structure: Chapter Problems

Question 6. Define the following terms:

(a) Atomic number (b) Mass number

(c) Nucleons (d) Valence shell Answer:

(a) Atomic number: Atomic number

refers to the number of protons

present in an atom. It is denoted by Z .

Example: An atom of oxygen contains

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8 proton Therefore its atomic number is 8.

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Datu Directions: read and answer each question in order to receive credit

(1) Indicate whether each of the following is a permissible set of quantum numbers. If the set is not permissible, state why it is not a.

2,40 $m=+1$ b. $n.5,3, m-3$ cn $4,1= 3, m2$

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d. $n=0, l=0, m=+1$ (2) Indicate whether each of the following is a permissible set of quantum numbers.

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