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HBr 4 NH4OH - KAl (SO4)2 12H2O - Al (OH)3 (NH4)2SO4 - KOH 12 H2O When you balance the chemical equation, it's always a good idea, to check the final equation, to make sure it works. Check: Reduce the numbers of each type of atom. The total number of atoms in the balanced equation will be the same on both sides of the equation.

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To set up an equation in the Chemical Equations Gizmo, type the chemical formulas into the text boxes of the Gizmo. First, type in “H2+O2” in the Reactants box and “H2O” in the Products box. This...

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2 Gizmo Warm-up In a chemical reaction, reactants interact to form products. This process is summarized by a chemical equation. In the Balancing Chemical Equations Gizmo, look at the floating molecules below the initial reaction: H 2 + O 2 ? H 2 1. How many atoms are in a hydrogen molecule (H 2)? O. 2

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Balancing Chemical Equations Gizmo Answer Key

Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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Launch Gizmo Practice balancing chemical equations by changing the coefficients of reactants and products. As the equation is manipulated, the amount of each element is shown as individual atoms, histograms, or numerically. Molar masses of reactants and products can also be calculated and balanced to demonstrate conservation of mass.

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Gizmo Answer Key Chemical Equations

Which set of diagrams shows a balanced chemical equation? A. Diagram A B. Diagram B C. Diagram C D. Diagram D Explanation: In answer C, each side contains 4 atoms of hydrogen (H), 2 atoms of oxygen (O), and 3 atoms of sulfur (S). This is balanced because there is the same number of each type of atom on each side of the equation.

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